

Enantioselective Synthesis of a Bicyclic Ketal Induced by Chiral Sulfoxides : (-)-(1*R*, 3*R*, 5*S*)-endo-1,3-Dimethyl-2,9-Dioxabicyclo-[3,3,1]-Nonane.

Guy Solladié* and Nathalie Huser

Ecole Européenne des Hautes Etudes des Industries Chimiques (EHICS), Université Louis Pasteur, Laboratoire de
Stéréochimie associé au CNRS, 1 Rue Blaise Pascal, 67008 - Strasbourg, France.

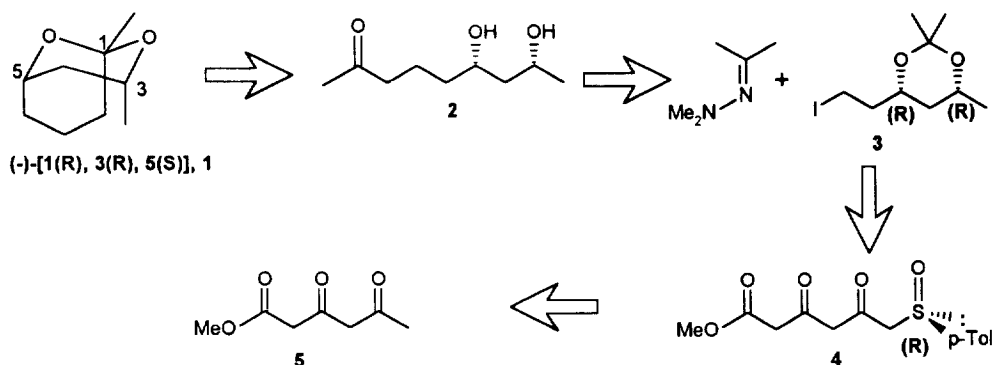
Abstract: The bicyclic ketal, (-)-(1*R*,3*R*,5*S*)-1,3-dimethyl-2,9-dioxabicyclo-[3,3,1]-nonane, was prepared by a short enantioselective reduction of an enantiomerically pure β,δ -diketosulfoxide.

Endo-1,3-dimethyl-2,9-dioxabicyclo [3,3,1]-nonane, **1**, is a host specific substance for the ambrosia beetle *Trypodendron lineatum* Oliver that infests the bark of the Norway Spruce. It was isolated and identified in 1976 by Heemann and Francke¹ from extracts of the bark of trees attacked by this beetle. However, neither the absolute configuration nor the specific rotation have not been reported. Since then, many stereoselective syntheses of the racemate² and both enantiomers³ have been published. Most of the preparations of the optically active stereomers started from the chiral pool or from optically active precursors obtained via enzymatic reactions⁴. There is until now only one report^{3j} describing an asymmetric synthesis of compound **1**, using a sultam to induce the chirality.

We report in this paper an efficient enantioselective synthesis of **1** induced by a chiral sulfoxide group. The strategy of the synthesis is shown in the retrosynthetic Scheme I. The dihydroxyketone **2**, precursor of the acetal **1** can be readily prepared by coupling the anion of the *N,N*-dimethyl hydrazone of acetone and the dihydroxyiodide **3**. Compound **3** contains a *syn*-diol functionality which can be obtained by reduction of a diketosulfoxide as we have shown in our previous work⁴. We reported⁵ also different ways to prepare enantiomerically pure diketosulfoxides. Following these results the diketoester **5** was the best precursor of **4**.

The diketoester **5** was obtained in one step from commercially available dehydroacetic acid by a known procedure⁶. Condensation of the trianion of **5**, prepared in THF with one equivalent of NaH and 2 equivalents of *t*-BuLi at 0°C to (-)-menthyl (*S*)-*p*-toluenesulfinate⁷ afforded in high yield the (*R*)-diketosulfoxide **4** (Scheme II). As expected⁴, only the δ -carbonyl was entirely enolized (one vinylic hydrogen giving a singlet at 5.65 ppm).

Scheme 1



Following our previous results⁴, reduction with DIBAL (2 eq. in THF at $-78^{\circ}C$) gave only one diastereomer of **6**⁸. The relatively low yield of isolated product is due to some decomposition of the product during the chromatographic purification. The absolute configuration $3(R),S(R)$ of compound **6** was deduced from our previous results⁴ and will be confirmed by correlation with the known compound **1**.

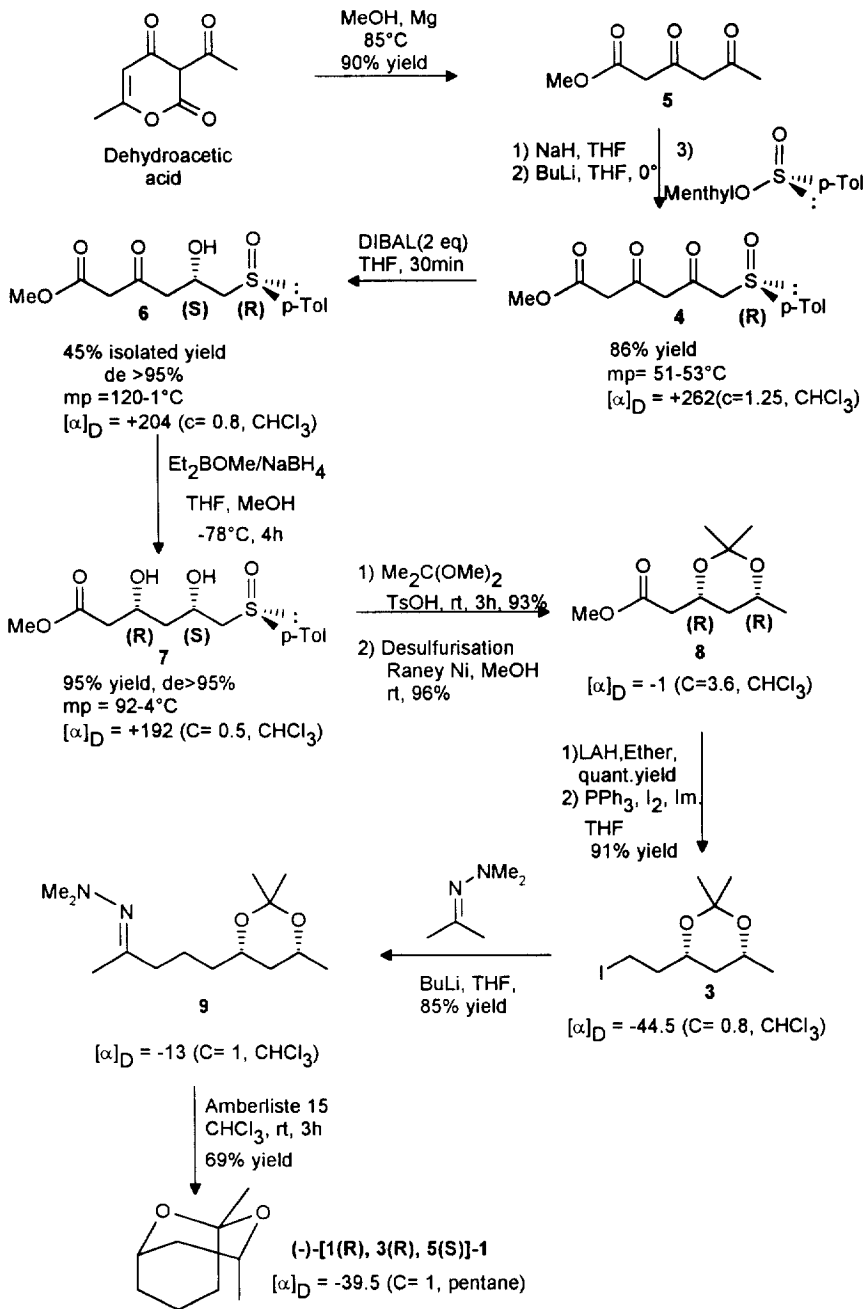
In the next step the δ -carbonyl was reduced to the syn diol **7** with $Et_2BOMe/NaBH_4$ ⁹ in a quantitative yield and $>95\%$ d.e. (only one stereomer was detected by 1H NMR¹⁰). Protection of the diol and desulfurization with Raney Nickel gave the ester **8** which was reduced to a primary alcohol and transformed into the corresponding iodide. Finally condensation with the anion of *N,N*-dimethyl hydrazone of acetone followed by hydrazone and acetonide hydrolysis with amberlist 15 afforded the bicyclic acetal **1** showing all the characteristics of the known product. The specific rotation of this highly volatile material, which was difficult to purify, was in the range of the literature values^{3d-g, 3j} (between -35.6 and -46.5).

This enantioselective synthesis of the bicyclic ketal **1** is one of the shortest ever reported. It could be very easily applied to the preparation of all the others stereoisomers just by changing the absolute configuration of the starting sulfoxide and either forming the *syn*- or *anti*-diol by known procedure^{4b}.

References and notes.

- 1) a) Heeman, V. ; Francke, W., *Naturwissenschaften*, **1976**, *63*, 344. b) Vité, J.P. ; Francke, W., *Naturwissenschaften*, **1976**, *63*, 550-555.
- 2) a) Kongkathip, B. ; Kongkathip, N., *Tetrahedron Lett.* **1984**, *25*, 2175-2176. b) Mohr, P. ; Tamm, C., *Tetrahedron Lett.* **1987**, *28*, 395-6. c) Adams, J. ; Frenette, R., *Tetrahedron Lett.* **1987**, *28*, 4773-4774. d) Funk, R.L. ; Bolton, G.L., *Tetrahedron Lett.* **1988**, *29*, 1111-1114.

Scheme II



- 3) a) Page, P.C.B. ; Rayner, C.M. ; Sutherland, I.O., *Tetrahedron Lett.* **1986**, *27*, 3535-3538 and *J. Chem. Soc. Perkin Trans I* **1990**, 2403-2408 : (+)-endo-(1S, 3S, 5R) via Sharpless epoxidation. b) Bernardi, R. ; Ghiringhelli, D., *J. Org. Chem.* **1987**, *52*, 5021-5022 : (+)-endo-(1S, 3S, 5R) and (+)-exo-(1S, 3R, 5R) via microbial reduction of carbonyl. c) Chikashita, H. ; Kittaka, E. ; Kimura, Y. ; Itoh, K., *Bull. Chem. Soc. Jpn* **1989**, *62*, 833-37 : (+)-endo-(1S, 3S, 5R) and (-)-exo-(1R, 3S, 5S) via Baker's yeast reduction of carbonyl. d) Redlich, H. ; Schneider, B. ; Francke, W. ; *Tetrahedron Lett.* **1980**, *21*, 3009-3012 and 3013-3016 ; Redlich, H. ; Schneider, B. ; Hoffmann, R.W. ; Geucke, K-J., *Liebigs Ann. Chem.* **1983**, 393-411 : the four isomers prepared from D- Glucose. e) Guindon, Y. ; St. Denis, Y. ; Daigneault, S. ; Morton, H.E., *Tetrahedron Lett.* **1986**, *27*, 1237-1240 and Mori, Y. ; Kuhara, M. ; Suzuki, M., *Chem. Pharm. Bull.* **1989**, *37*, 1078-1079 : (-)-endo-(1R, 3R, 5S) from (S)-malic acid. f) Masoni, C. ; Deschenaux, P.F. ; Kallimopoulos, T. ; Jacot-Guillarmot, A., *Helv. Chim. Acta* **1989**, *72*, 1284-1287 and Mohr, P. ; *Tetrahedron Lett.* **1991**, *32*, 2223-224 : (-)-endo-(1R, 3R, 5S) from ethyl (R)-3-hydroxybutyrate. g) Nakata, T. ; Nagao, S. ; Takao, S. ; Tanaka, T. ; Oishi, T., *Tetrahedron Lett.* **1985**, *26*, 73-74 : (-)-endo (1R, 3R, 5S) from (R)- 1,3-butanediol. h) Dauphin, G. ; Fauve, A. ; Veschambre, H., *J. Org. Chem.* **1989**, *54*, 2238-2242 : (+)-endo (1S, 3S, 5R) via microbial reduction of 1,3-diketones. i) Bloch, R. ; Seck, M., *Tetrahedron Asym.* **1990**, *1*, 855-856 : (-)-endo-(1R, 3R, 5S) via enzymatic hydrolysis of a racemic lactol. j) Curran, D.P. ; Heffner, T.A., *J. Org. Chem.* **1990**, *55*, 4585-4595 : (-)-endo-(1R, 3R, 5S) via asymmetric synthesis with chiral sultam.
- 4) a) Solladié, G. ; Ghiatou, N., *Tetrahedron Lett.* **1992**, *33*, 1605-1608. b) Solladié, G. ; Dominguez, C., *J. Org. Chem.* **1994**, *59*, 3898-3901.
- 5) a) Solladié, G. ; Ghiatou, N., *Tetrahedron Asym.* **1992**, *3*, 33. b) Solladié, G. ; Bauder, C. ; Rossi, L. Submitted to *J. Org. Chem.* **1995**, in press..
- 6) Batelaan, J. G., *Synth. Comm.* **1976**, *6*, 81
- 7) Solladié, G. ; Hutt, J. ; Girardin, A., *Synthesis* **1987**, 713
- 8) ¹NMR of **6** (CDCl₃, 200 MHz): δ : 2.42 (s, 3H), 2.79 (d, 2H, J=6.5 Hz), 2.90 (AB part of ABX, 2H, J_{AB}=13.5 Hz, J_{AX}=2.5 Hz, J_{BX}= 9.5 Hz, Δv=59 Hz), 3.48 (s, 2H), 3.71 (s, 3H), 4.22 (d, 1H, J=3.5 Hz), 4.62 (m, 1H), 7.41[(AB)₂, 4H, J_{AB}=8.0 Hz, Δv=35 Hz].
- 9) Chen, K.M. ; Hardtmann, G.E. ; Prasad, K. ; Repic, O. ; Shapiro, M.J., *Tetrahedron Lett.* **1987**, *28*, 155.
- 10) ¹NMR of **7** (CDCl₃, 200 MHz): δ : 1.53-1.76 (m, 2H), 2.42 (s, 3H), 2.45 (m, 2H), 2.85 (AB part of ABX, 2H, J_{AB}=13.5 Hz, J_{AX}= 2.5 Hz, J_{BX}=9.5 Hz, Δv= 80 Hz), 3.69 (s, 3H), 3.88 (d, 1H, J=2.5 Hz), 4.30 (m, 1H), 4.47 (m, 1H), 4.67 (d, 1H, J=1.5 Hz), 7.43 [(AB)₂, 4H, J_{AB}= 8.0 Hz, Δv= 34 Hz].